

# Development of a Novel Dry Chemical Uranium/Molybdenum Separation: Research for a Future Efficient Mo-99 Extraction Process

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# On Current Waste Management and the Conversion of HEU to LEU Targets

## Waste Management Considerations:<sup>[1]</sup>

- Conversion of HEU to LEU targets can generate as much as 200% more radioactive waste
- Production of 10,000 6-day Ci/week using LEU targets is expected to produce 15,000 liters per year of intermediate level wastes (ILWs).
  - After cementation, volume can increase to 375,000 liters

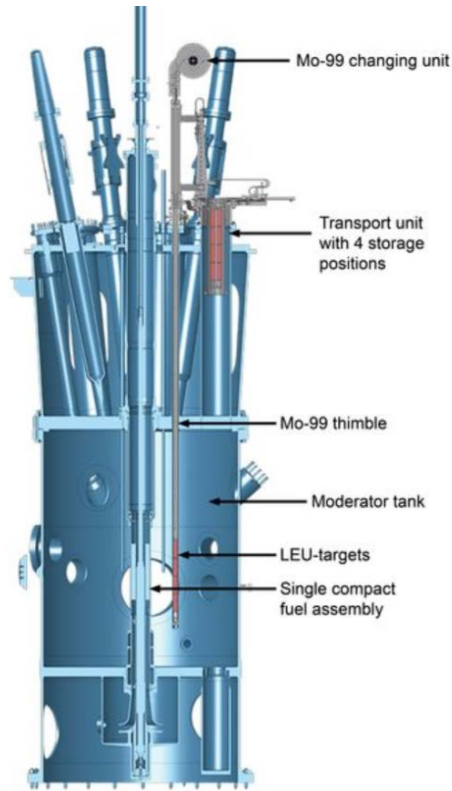
[1] S.K. Lee, G.J. Beyer, J.S. Lee. Nuclear Engineering and Technology. 48 (2016) 613-623.

## Typical Molybdenum-99 Production Scheme



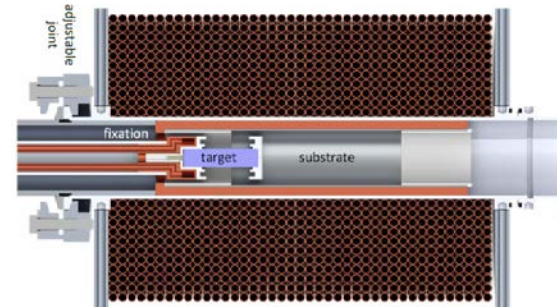
# Molybdenum-99 at FRM II

## Production

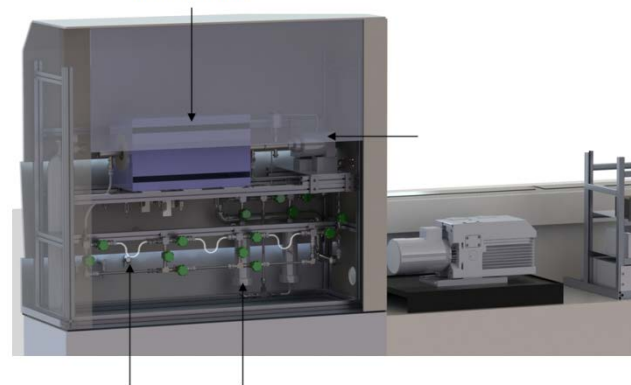


Irradiation Facility

## Research



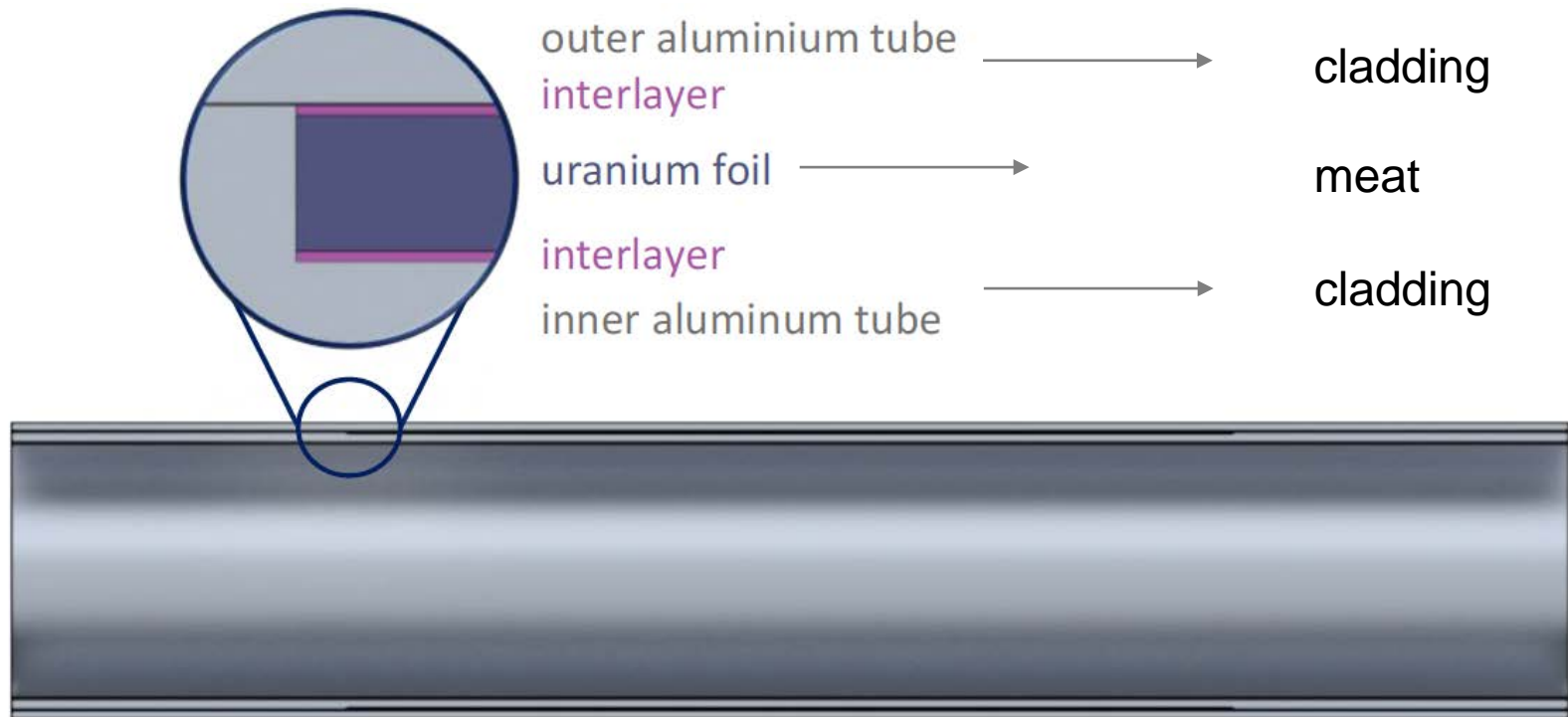
Target Design



Extraction Process



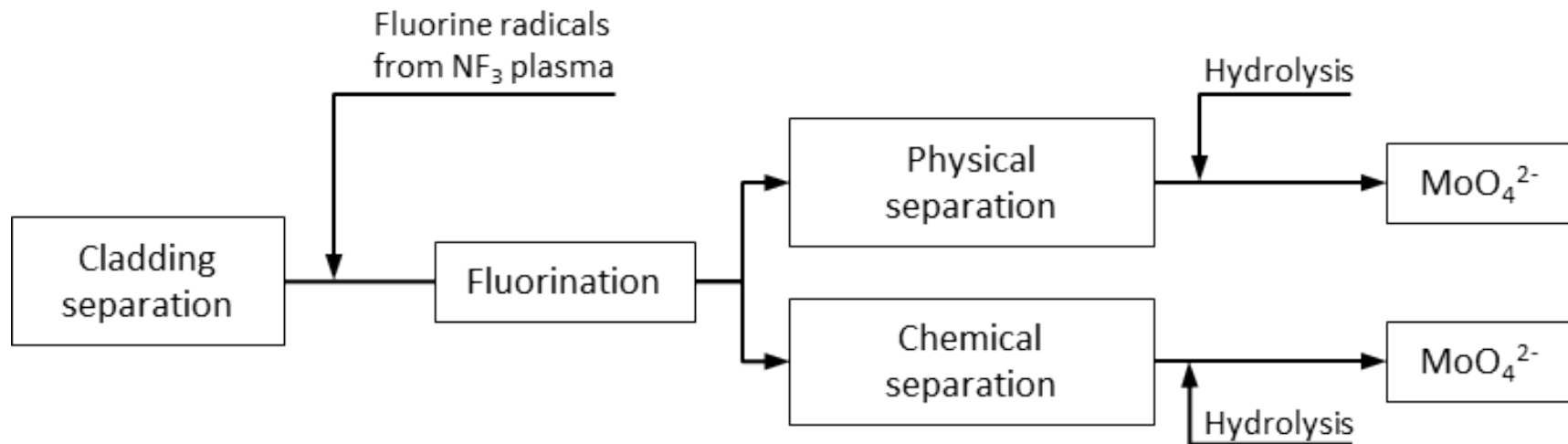
# Cylindrical Target Design



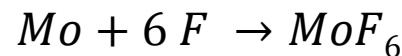
With cylindrical design, aluminum cladding can be mechanically separated before dissolution.

- This will lead to decreases in liquid waste production

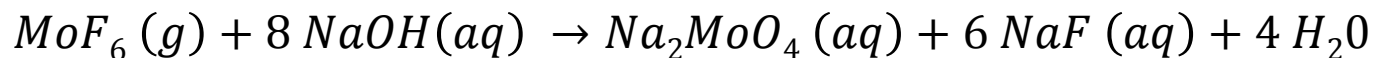
# Proposed Dry Chemical, Plasma-Aided Fluorination Process



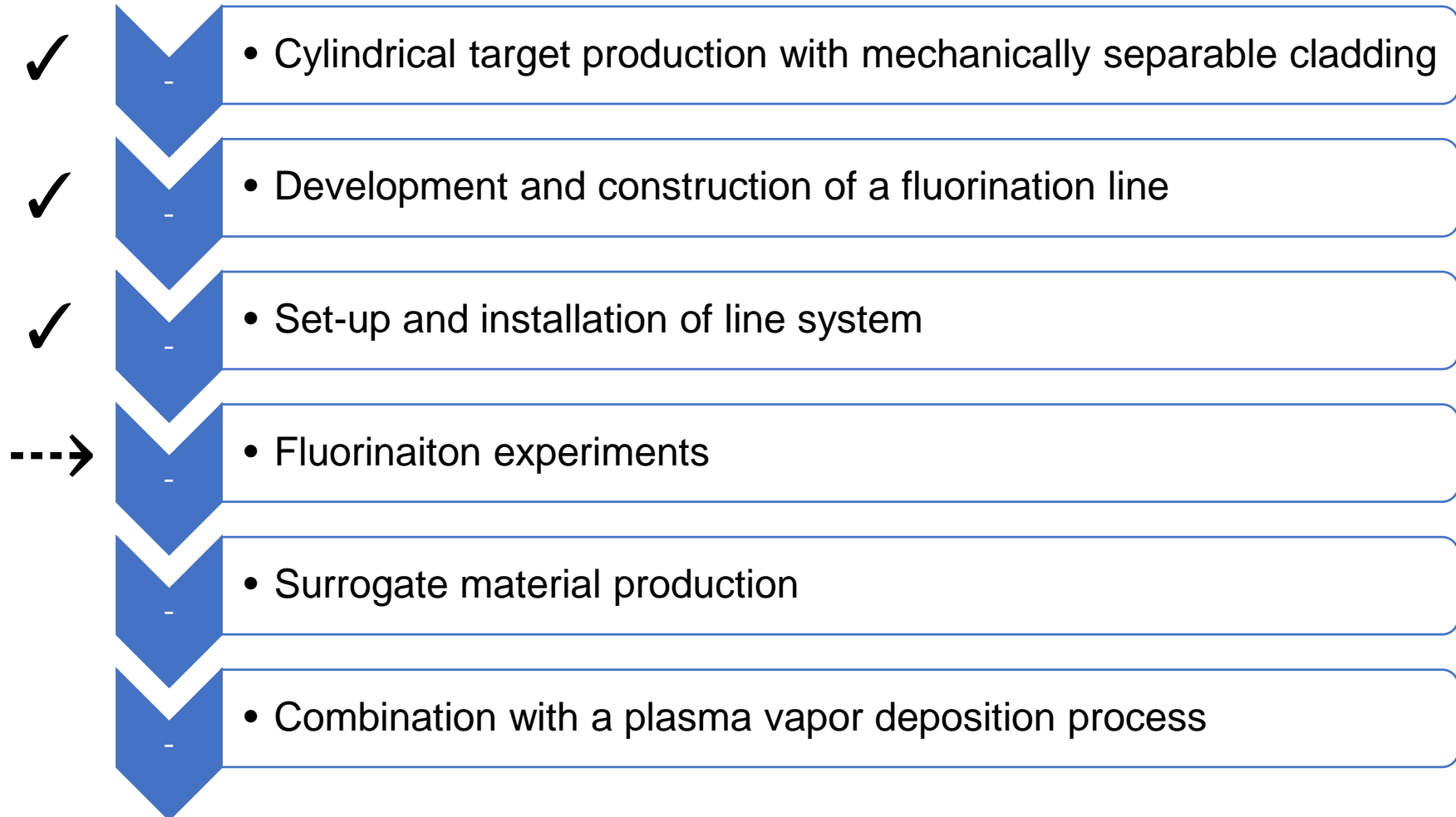
- Uranium and fission products fluorinated to higher fluorides with aid of plasma



- After separation, molybdenum hydrolyzed to  $MoO_4^{2-}$  using sodium hydroxide



# The Road to Dry-Chemical Separation



# Production of a Surrogate Material

Element	Mass [mg]	m.p. [°C]	b.p. [°C]
Xe	35,5	-111,7	-108
Zr	28,5	1857	4409
Ce	25,2	798	3426
Mo	17,5	2623	4639
Nd	16,7	1024	3100
Ba	15,5	727	1640
Ru	14,3	2334	4150
Sr	11,8	777	1382
Cs	11,2	28,5	671
La	9,3	920	3457
Pr	6,47	930,8	3512
Y	5,03	1522	2930
Te	4,87	449,5	987,8
Tc	3,61	2204	4265
Pu	3,32	639,4	3230
I	3,25	113,7	184
Kr	3,06	-157,4	-152
Rb	3,04	39,3	688
Sm	2,23	1072	1900
Np	1,57	639	3902
Pm	1,29	1080	3000
Pd	1,10	1555	2960

- Inactive (excluding uranium) surrogate material containing most important elements
- About 10 fission products to be taken into consideration
- Investigate the behavior of surrogate materials in plasma line
- Process surrogate materials through the most promising physical and chemical separation techniques

## Chemical Investigations after Fluorination

- Higher fluorides of uranium and molybdenum expected from plasma-induced fluorination.

HF																		He
							MnF <sub>3</sub> , MnF <sub>4</sub>	FeF <sub>3</sub>	CoF <sub>3</sub> , CoF <sub>2</sub>	NiF <sub>2</sub>	CuF <sub>2</sub>	ZnF <sub>2</sub>	GaF <sub>3</sub>	GeF <sub>4</sub>	AsF <sub>3</sub> , AsF <sub>5</sub>	SeF <sub>4</sub> , SeF <sub>6</sub>	BrF <sub>3</sub>	Kr, KrF <sub>2</sub>
RbF	SrF <sub>2</sub>	YF <sub>3</sub>	ZrF <sub>4</sub>	NbF <sub>5</sub>	MoF <sub>6</sub>	TcF <sub>6</sub>	RuF <sub>6</sub> , RuF <sub>3</sub>	RhF <sub>6</sub> , RhF <sub>3</sub>	PdF <sub>2</sub>	AgF, AgF <sub>2</sub>	CdF <sub>2</sub>	InF <sub>3</sub>	SnF <sub>4</sub>	SbF <sub>5</sub>	TeF <sub>6</sub>	IF <sub>5</sub> , IF <sub>7</sub>	Xe, XeF <sub>2,4,6</sub>	
CsF	BaF <sub>2</sub>	LaF <sub>3</sub>																

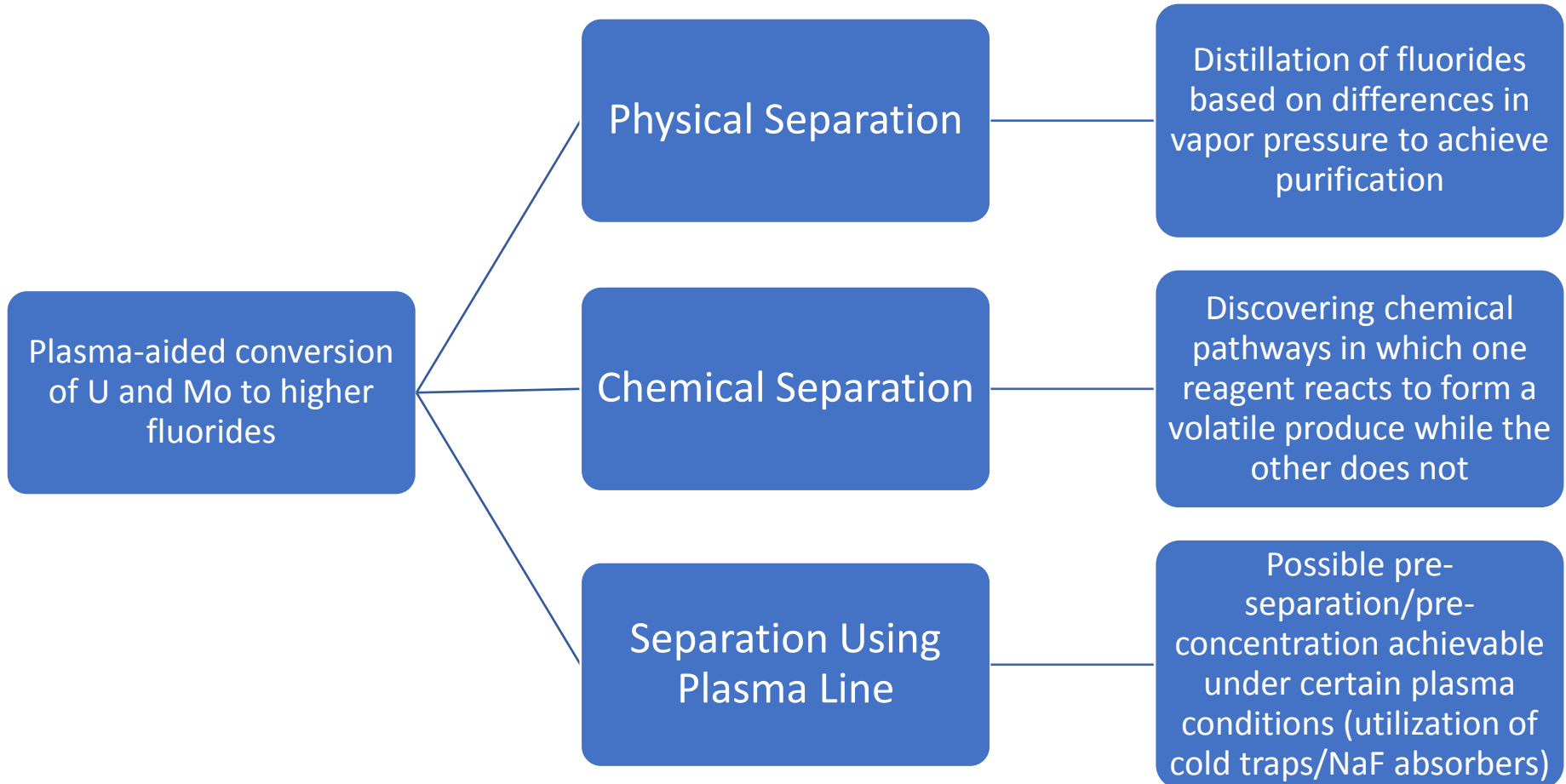
Lanthanoids	CeF <sub>4</sub>	PrF <sub>3</sub>	NdF <sub>3</sub>	PmF <sub>3</sub>	SmF <sub>3</sub>	EuF <sub>3</sub>	GdF <sub>3</sub>	TbF <sub>3</sub>	DyF <sub>3</sub>	HoF <sub>3</sub>	ErF <sub>3</sub>			
Actinoids	ThF <sub>4</sub>	PaF <sub>3</sub>	UF <sub>6</sub>	NpF <sub>6</sub>	PuF <sub>6</sub>	AmF <sub>3</sub>	CmF <sub>3</sub>							

- Major volatile fluorides: UF<sub>6</sub>, MoF<sub>6</sub>, TeF<sub>6</sub>, TcF<sub>6</sub>, PuF<sub>6</sub>, IF<sub>5</sub>, IF<sub>7</sub>, NpF<sub>6</sub>
- Current focus on the separation of molybdenum from uranium



## Proposed Method of Fluoride Separation

- All proposed separation methods are based on dry-chemical techniques. If solvents are considered, only those with very low boiling points will be utilized.



# Wet Chemical Processing versus Dry Chemical Processing of Molybdenum-99

	Wet Processing	Dry Processing
Advantages	<ul style="list-style-type: none"> <li>▪ Produces large quantities of <math>\text{MoO}_4^{2-}</math> with high specific activity</li> <li>▪ Highly developed processes with efficiencies up to 90%</li> </ul>	<ul style="list-style-type: none"> <li>▪ Simple production of targets</li> <li>▪ Requires processing of only irradiated uranium foil</li> <li>▪ Significant reduction of liquid radioactive waste</li> <li>▪ Possible process efficiencies &gt; 90%</li> </ul>
Disadvantages	<ul style="list-style-type: none"> <li>▪ Produces high volumes of liquid radioactive waste</li> </ul>	<ul style="list-style-type: none"> <li>▪ Carrier-free separation yet to be evaluated</li> </ul>

# Questions?

